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Our results confirm and build upon the hypothesis that high dilutions of biologically active compounds may indeed have an effect which mimics that of lower dilutions/higher doses. So far there is no satisfactory or uniting theoretical explanation for these observations, but recent evidence seems to point to organisation of the solvent water on a mesoscopic scale: the nanoheterogenous structure of water can be determined by interactive phenomena such as coherence [1;2], epitaxy [3], temperature-pressure processes during strong agitation, and formation of colloidal nanobubbles containing gaseous inclusions of oxygen, nitrogen, carbon dioxide, silica, and possibly the remedy source material [4-6]. So far, nuclear magnetic resonance (NMR) studies of homeopathic solutions have yielded controversial data, but recently high histamine dilutions (beyond 10^{-20} mol/L) were distinguished from water solvents on the basis of their NMR relaxation rates [4;7;8]; moreover, after heating of the histamine samples, all the relaxation observed as a function of dilution vanished. It was suggested that histamine molecules might act as nucleation centres, amplifying the formation of stable supramolecular structures, involving nanobubbles of atmospheric gases and highly ordered water around them. In the future, the possible existence and the nature of clatrate-like hydrate nanostructures formed during the dilution and succussion process might be explained by cluster science, in which different geometrical structures of clusters composed of the same chemical species may differ in their chemical reactivity [9]. These unusual properties of high dilutions, which merit further investigation, are potentially relevant not just to homeopathic pharmaceutical practice, but also to basic research into cell sensitivity to regulation.

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